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CROSS-REFERENCE TO RELATED APPLICATIONS

This application relates to Applicants' concurrently filed U.S. applications Attorney Docket No. PET-2133, "Catalyst Comprising at Least One Zeolite Chosen

5 from ZBM-30, SM-48 EU-2 and EU-11 and at Least one Y Zeolite and Process of Hydroconversion of Hydrocarbon Charges Using Such a Catalyst" (French priority 03/03.529 filed March 24, 2003) and Attorney Docket No. PET-2135, "Catalyst and Its Use for Improving the Pour Point of Hydrocarbon Charges" (French priority 03/03.531 filed March 24, 2003).

10 The present invention relates to a process for improving the pour point of charges containing long (more than 10 carbon atoms) linear and/or slightly branched paraffins, in particular in order to convert, with a good yield, charges having high pour points to at least one cut having a low pour point and a high viscosity index for the oil bases.

15 A subject of the present invention is also a catalyst comprising at least one ZBM-30 zeolite, synthesized in the presence of a particular structuring agent such as triethylenetetramine, at least one porous mineral matrix, at least one hydro-dehydrogenating element, preferably chosen from the elements of Group VIB and Group VIII of the periodic table, and its use for the conversion of hydrocarbons and 20 the process for reducing the pour point with this catalyst.

Prior art

High-quality lubricants are of fundamental importance for the satisfactory operation of modern machinery, cars and lorries. However, the quantity of untreated 25 paraffins originating directly from petroleum, and having suitable properties for constituting good lubricants is very small compared with the growing demand in this sector.

The treatment of heavy petroleum fractions with high linear or slightly branched paraffins contents is necessary in order to obtain high-grade base oils with 30 the best possible yields, by an operation aimed at eliminating the linear or very slightly branched paraffins, from charges which will then be used as base oils or as kerosene or jet fuel.

In fact paraffins with a high molecular weight, which are linear or very slightly branched and which are present in oils or in the kerosene or jet fuel lead to

high pour points and therefore to solidification phenomena for low-temperature uses. In order to reduce the pour points, these linear, unbranched or very slightly branched paraffins must be entirely or partially eliminated.

5 This operation can be carried out by extraction using solvents such as propane or methyl ethyl ketone, which is called propane or methyl ethyl ketone (MEK) dewaxing. However, these techniques are expensive, lengthy and not always easy to implement.

10 Another means is selective cracking of the longest linear paraffin chains which leads to the formation of compounds with a lower molecular weight, some of which can be eliminated by distillation.

15 Given their form selectivity, zeolites are among the most widely-used catalysts. The idea underlying their use is that zeolitic structures exist, the pore openings of which are such that they allow the entry of the long linear or very slightly branched paraffins into their micropores, but exclude branched paraffins, naphthenes and aromatics. This phenomenon thus leads to a selective cracking of the linear or very slightly branched paraffins.

20 Catalysts based on zeolites having intermediate pore sizes such as ZSM-5, ZSM-11, ZSM-12, ZSM-22, ZSM-23, ZSM-35 and ZSM-38 have been described for their use in these processes in particular in the patents US-A-3 894 938; US-A-4 176 050; US-A-4 181 598; US-A-4 222 855; US-A-4 229 282 and US-A-4 247 388.

Mixtures of zeolites with large pores and these zeolites with intermediate pores which can be used in a dewaxing process are described in Patent Application WO-A-02/08 8279.

25 Moreover, it was found that processes using these zeolites (ZSM-5, ZSM-11, ZSM-12, ZSM-22, ZSM-23, ZSM-35 and ZSM-38) make it possible to obtain oils by cracking of charges containing quantities of linear or very slightly branched paraffins of less than 50% by weight. However, for charges containing higher quantities of these compounds it transpired that their cracking leads to the formation of large quantities of products with lower molecular weights, such as butane, propane, ethane 30 and methane, which considerably reduces the yield of sought products.

In order to remedy these drawbacks, the applicant concentrated his research on the developing of catalysts comprising at least one ZSM-30 zeolite, at least one hydro-dehydrogenating element, preferably chosen from the elements of Group VIB and Group VIII of the periodic table. The Applicant then surprisingly discovered that

the use of a catalyst comprising at least one zeolite ZSM-30 synthesized with a particular structuring agent such as triethylenetetramine makes it possible to reduce the pour point of the charge while still obtaining a high viscosity index (VI) and maintaining a good yield of desired products. Unexpectedly, the catalyst prepared 5 with the ZSM-30 zeolite synthesized in the presence of the triethylenetetramine structuring agent has a greater de-waxing activity and selectivity (improvement of the pour point) than do the solids based on ZSM-30s synthesized via other structuring agents and the zeolite-based catalytic formulae known from the prior art.

10 The present invention proposes a catalytic process for reducing the pour point, based on such catalysts.

Subject of the invention

15 A first subject of the invention is a catalyst comprising at least one ZBM-30 zeolite synthesized with a particular structuring agent such as triethylenetetramine, at least one hydro-dehydrogenating element, preferably chosen from the elements of Group VIB and Group VIII of the periodic table.

20 Another subject of the invention is a process for improving the pour point of a hydrocarbon charge, in which the charge to be treated is brought into contact with a catalyst comprising at least one zeolite synthesized with a particular structuring agent such as triethylenetetramine, at least one hydro-dehydrogenating element, preferably chosen from the elements of Group VIB and Group VIII of the periodic table, at least one porous mineral matrix. This dewaxing process is carried out at a temperature between 200 and 450°C, a pressure between 0.1 and 25 MPa and an hourly volume rate between 0.05 and 30h⁻¹ in the presence of hydrogen at a level of 50 to 2000 25 normal litres of hydrogen per litre of charge (Nl/l).

The synthesis of the ZBM-30 zeolite is described in the patent EP-A-46504.

Unexpectedly, said catalyst has a greater dewaxing activity and selectivity (improvement of the pour point) than the zeolites-based catalytic formulae (molecular sieve) known from the prior art.

30 Advantageously, this process makes it possible to convert a charge having a high pour point to a product having a lower pour point. It is therefore applicable for reducing the pour point of the gasoils for example. It also makes it possible to reduce the pour point of heavier charges in order to obtain oil bases having good cold-weather properties and a high viscosity index.

The charges which can be treated according to the process of the invention are advantageously fractions having relatively high pour points, the value of which it is wished to reduce.

Typical charges which can be treated advantageously according to the invention generally have a pour point above 0°C. The products resulting from the treatment according to the process have pour points below 0°C and preferably below approximately -10°C.

The charges which can be treated according to the process of the invention contain paraffins, olefins, naphthenes, aromatics and also possibly organic nitrogen, sulphur and oxygen compounds.

The paraffins resulting from the Fischer-Tropsch process are excluded.

The use of the catalyst according to the invention under the conditions described above allows, in particular, the production of products having a low pour point with good yields, and a high viscosity index in the case of the heaviest fractions which are treated with a view to producing oil bases.

Detailed description of the invention

The catalyst according to the invention comprises at least one ZBM-30 zeolite synthesized with a particular structuring agent such as triethylenetetramine, at least one hydro-dehydrogenating element, preferably chosen from the elements of Group VIB and Group VIII (preferably the noble metals of Group VIII) of the periodic table and at least one porous mineral matrix.

The synthesis of the ZBM-30 zeolite is described in the patent EP-A-46504.

The ZBM-30 zeolite is synthesized according to the methods described in the patent EP-A-46504 according to the procedure using the structuring agent triethylenetetramine.

The overall Si/Al ratio of the zeolites contained in the composition of the catalysts according to the invention as well as the chemical composition of the samples are determined by X-ray fluorescence and atomic absorption.

The Si/Al ratios of the zeolites described above are those obtained during the synthesis according to the procedures described in the various documents cited or obtained after post-synthesis dealumination treatments well known to a person skilled in the art, such as, non-exhaustively, hydrothermal treatments followed or not

followed by acid attacks or also direct acid attacks by solutions of mineral or organic acids.

The zeolites contained in the composition of the catalyst according to the invention are calcined and exchanged by at least one treatment with a solution of at 5 least one ammonium salt in order to obtain the ammonium form of the zeolites which, once calcined, lead to the hydrogen form of the zeolites.

The zeolites contained in the composition of the catalyst according to the invention are at least in part, preferably practically totally, in acid form, i.e. in the 10 hydrogen form (H^+). The Na/T atomic ratio is generally less than 10% and preferably less than 5% and even more preferably less than 1%.

On the other hand, the catalyst forming the subject of the invention contains at least one hydro-dehydrogenating element, preferably chosen from the elements of Group VIB and Group VIII (i.e. metal or compound) of the periodic table and at least one porous mineral matrix.

15 In the case where the element is at least one metal of Group VIII, preferably when it is a noble metal and advantageously a noble metal chosen from the group formed by platinum and palladium, it can be introduced onto the zeolites for example by dry impregnation, by ion exchange or any other process known to a person skilled in the art, or indeed it can be introduced onto the matrix.

20 According to a first variant, prior to its shaping, the zeolite previously described is subjected to the application of at least one metal of Group VIII, preferably chosen from the group formed by platinum and palladium.

The zeolite is then shaped by any technique known to a person skilled in the art. It can in particular be mixed with a matrix, generally amorphous, for example a 25 moist alumina gel powder. The mixture is then shaped, for example by extrusion through a die.

The shaping can be carried out with matrices other than alumina, such as for example magnesia, amorphous silica-aluminas, natural clays (kaolin, bentonite, sepiolite, attapulgite), silica, titanium oxide, boron oxide, zirconium, aluminium 30 phosphates, titanium phosphates, zirconium phosphates, charcoal and their mixtures. It is preferred to use matrices containing alumina, in all its forms known to a person skilled in the art, and still more preferably aluminas, for example gamma alumina. Techniques other than extrusion, such as pelleting and dragée production, can be used.

Mixtures of alumina and silica, mixtures of alumina and silica-alumina can also advantageously be used.

The catalysts obtained by the present invention are shaped in the form of grains of different forms and dimensions. They are generally used in the form of 5 cylindrical or polylobate extrudates such as straight or twisted bilobates, trilobates, polylobates, but can optionally be produced and used in the form of crushed powder, tablets, rings, beads, coils.

After the shaping stage, the product obtained is subjected to a drying stage, then to a calcination stage.

10 In the case where the hydrogenating metal belongs to Group VIII, and is preferably Pt and/or Pd, it can also and advantageously be applied to the support after the shaping of the zeolite which is free of metals, using any process known to a person skilled in the art and allowing the application of the metal to the molecular sieve. In this case the support is obtained in a manner analogous to that described previously.

15 In the remainder of the text, the term support will be used to describe the zeolite (free of metals), plus the matrix after shaping, drying and calcination, for example as obtained previously.

20 In order to apply the metal to the zeolite, it is possible to use the cation exchange technique with competition where the competitor is preferably ammonium nitrate, the competition ratio being at least equal to approximately 20 and advantageously approximately 30 to 200. In the case of platinum or palladium, a 25 tetramine complex of platinum or a tetramine complex of palladium is customarily used: these latter will then be applied practically totally to the zeolite. This cation exchange technique can also be used in order to apply the metal directly to the molecular sieve powder, before its optional mixing with a matrix.

30 The application of the metal (or metals) of Group VIII is generally followed by calcination under air or oxygen, usually between 300 and 600°C for 0.5 to 10 hours, preferably between 350 and 550°C over 1 to 4 hours. Reduction under hydrogen can then be carried out, generally at a temperature between 300 and 600°C for 1 to 10 hours, preferably operating between 350 and 550°C for 2 to 5 hours.

It is also possible to apply the platinum and/or palladium no longer directly to the zeolite, but to the matrix (for example the aluminium binder) of the support, before or after the shaping stage, implementing an anion exchange with hexachloroplatinic acid, hexachloropalladic acid and/or palladium chloride in the

presence of a competing agent, for example hydrochloric acid. Generally after the application of platinum and/or palladium, the catalyst is, as previously, subjected to calcination then reduced under hydrogen as indicated above.

5 The support of the catalyst of the present invention generally covers the following matrix and zeolites contents:

- 5 to 95% by weight, preferably 10 to 90% by weight, more preferably 15 to 85% by weight and very preferably 20 to 80% by weight of ZBM-30 zeolite,
- 5 to 95%, preferably 10 to 90% by weight, more preferably 15 to 85% and very preferably 20 to 80% by weight of at least one amorphous or poorly 10 crystallized oxide-type porous mineral matrix.

The noble metal(s) content thus optionally introduced, expressed as a percentage by weight relative to the total mass of the catalyst, is generally below 5%, preferably below 3%, still more preferably below 2% and generally below 1% by weight.

15 In the case where the catalyst comprises a hydrogenating metal of Group VIII, preferably a noble metal and advantageously platinum and/or palladium, the catalyst is generally reduced in the reactor in the presence of hydrogen and under conditions well known to a person skilled in the art.

20 In the case where the hydrogenating metal is not a noble metal, the elements of Group VIB and Group VIII optionally introduced into the catalyst according to the invention can be present totally or partially in metal and/or oxide and/or sulphide form.

Among the elements of Group VIB, molybdenum and tungsten are preferred.

25 The sources of elements of Group VIB which can be used are well known to a person skilled in the art. For example, among the sources of molybdenum and tungsten, it is possible to use oxides and hydroxides, molybdic and tungstic acids and their salts, in particular ammonium salts such as ammonium molybdate, ammonium heptamolybdate, ammonium tungstate, phosphomolybdic acid, phosphotungstic acid and their salts. Preferably ammonium oxides and salts are used, such as ammonium 30 molybdate, ammonium heptamolybdate and ammonium tungstate.

The catalyst of the present invention can contain a non-noble metal of Group VIII and preferably cobalt and nickel. Advantageously, the following combinations of the non-noble elements of Groups VI and VIII are used: nickel-molybdenum, cobalt-molybdenum, iron-molybdenum, iron-tungsten, nickel-tungsten, cobalt-tungsten, the

preferred combinations are: nickel-molybdenum, nickel-tungsten. It is also possible to use combinations of three metals for example nickel-cobalt-molybdenum.

The sources of elements of Group VIII which can be used are well known to a person skilled in the art. For example, nitrates, sulphates, phosphates, halides will be 5 used, for example chlorides, bromides and fluorides, carboxylates for example acetates and carbonates.

When the hydrogenating function is provided by a non-noble metal of Group VIII or a combination of a non-noble metal of Group VIII and a metal of Group VIB,

- the composition of the support constituted by at least one matrix and zeolites 10 described in the invention is the same as that described previously and,
- the content by weight of at least one element chosen from the non-noble elements of Group VIB and Group VIII is between 0.1 and 60%, preferably between 1 and 50% and still more preferably between 2 and 40%.

Generally, in order to conclude the preparation of the catalyst, the moist solid 15 is left to rest under a moist atmosphere at a temperature between 10 and 80°C, then the moist solid obtained is dried at a temperature between 60 and 150°C, and finally the solid obtained is calcined at a temperature between 150 and 800°C, generally between 250 and 600°C.

The catalysts of the present invention can also be subjected to a sulphurization 20 treatment making it possible to convert, at least in part, the metal elements to sulphur before they are brought into contact with a charge to be treated. This activation treatment by sulphurizing is well known to a person skilled in the art and can be carried out by any method already described in the literature.

In the case of the non-noble metals, a standard sulphurizing treatment well 25 known to a person skilled in the art consists of heating in the presence or under flow of a hydrogen/hydrogen sulphide mixture or also under pure hydrogen sulphide, at a temperature between 150 and 800°C, preferably between 250 and 600°C, generally in a crossed-bed reaction zone.

30 Charges

The process makes it possible to convert a charge having a high pour point to a product having a lower pour point. It is therefore applicable for reducing the pour point of the gasoils for example. It also makes it possible to reduce the pour point of

heavier charges in order to obtain oil bases having good cold properties and a high viscosity index.

The hydrocarbon charges treated according to the process of the invention are advantageously fractions having relatively high pour points the value of which it is
5 wished to reduce.

Typical charges which can be treated advantageously according to the invention generally have a pour point above 0°C. The products resulting from the treatment according to the process have pour points below 0°C and preferably below approximately -10°C.

10 These charges preferably contain at least 20% by volume of compounds boiling above 340°C, preferably at least 350°C and advantageously at least 380°C. In the latter example, this does not mean that the boiling point is 380°C and above, but 380°C or above.

15 The process according to the invention is particularly suitable for treating paraffin distillates such as middle distillates which include gasoils, kerosenes, jet fuels, vacuum distillates and any other fractions the pour point and viscosity of which must be adapted in order to come within the scope of the specifications.

20 The charges which can be treated according to the process of the invention contain paraffins, olefins, naphthenes, aromatics and also possibly organic nitrogen, sulphur and oxygen compounds.

25 The charge can also be for example distillates resulting from the direct distillation of crude oil or conversion units such as FCCs, cokers or viscoreducers, or from aromatics extraction units, or from the desulphurization or hydroconversion of ATRs (atmospheric residues) and/or VRs (vacuum residues), or also the charge can be a deasphalting oil, or also a hydrocracking residue for example resulting from a hydrocracking stage, a vacuum distillate having undergone a hydrorefining stage, and more generally bases for lubricating oils, or also poly-alpha-olefins with a high pour point or any mixture of the abovementioned charges.

Fischer-Tropsch paraffins are excluded.

30 The charge to be treated preferably has a nitrogen compounds content below approximately 200 ppm by weight and preferably below 100 ppm by weight. The sulphur content is below 1000 ppm by weight, preferably below 500 ppm by weight, and still more preferably below 200 ppm by weight. The metals content of the charge,

such as Ni or V, is extremely small, i.e. below 50 ppm by weight, preferably below 10 ppm by weight and still more preferably below 2 ppm by weight.

The use of the catalyst according to the invention under the conditions described below makes it possible in particular to produce products with a low pour point with good yields, and with a high viscosity index in the case of the heaviest fractions which are treated with a view to producing oil bases.

Operating conditions

The operating conditions under which the catalytic dewaxing process of the invention is carried out are the following:

- the reaction temperature is between 200 and 450°C and preferably between 200 and 430°C, advantageously 250-420°C;
- the pressure is between 0.1 and 25 MPa and preferably between approximately 1 and 20 MPa.
- the hourly volume rate (hvr expressed as volume of charge injected per volume unit of catalyst per hour) is between approximately 0.05 and approximately 30 and preferably between approximately 0.1 and approximately 20 h^{-1} and still more preferably between approximately 0.1 and approximately 10 h^{-1} .

The contact between the charge and the catalyst is carried out in the presence of hydrogen. The level of hydrogen used and expressed in litres of hydrogen per litre of charge is between 50 and approximately 2000 litres of hydrogen per litre of charge and preferably between 100 and 1500 normal litres of hydrogen per litre of charge.

Embodiment

The catalytic dewaxing process according to the invention can be preceded by a hydroisomerization-hydroconversion stage in the presence of a catalyst containing at least one noble metal applied to an amorphous acid support.

This hydroisomerization-hydroconversion stage is optionally preceded by a hydrorefining stage, which hydrorefining stage can be followed by an intermediate separation.

The hydroisomerization-hydroconversion stage takes place in the presence of hydrogen and in the presence of a bifunctional catalyst comprising an amorphous acid support (preferably an amorphous silica-alumina) and a hydro-dehydrogenating metal function provided by at least one noble metal of Group VIII.

The support is called amorphous, i.e. having no molecular sieve, and in particular zeolite, as well as the catalyst. The amorphous acid support is advantageously an amorphous silica-alumina, but other supports can be used. When it is a silica-alumina, the catalyst, generally, does not contain an added halogen, other 5 that that which could be introduced for the impregnation, of the noble metal for example. The silica-alumina can be obtained by any synthesis technique known to a person skilled in the art such as the techniques of co-precipitation, cogelation, etc.

During the hydroisomerization-hydroconversion stage, the molecules of the charge to be treated, for example n-paraffins, in the presence of a bifunctional catalyst 10 undergo an isomerization then optionally a hydrocracking in order to lead respectively to the formation of isoparaffins and lighter cracking products such as gasoils and kerosene. The conversion of the products having boiling points above or equal to the initial boiling point of the charge which is at least 340°C, or even 370°C or, still better, at least 380°C, to products with boiling points below the initial boiling 15 temperature of the charge generally varies between 5 and 90%, preferably between 5 and 80%, but is generally preferably below 80% and still better below 60%.

The characteristics of the hydroisomerization-hydroconversion catalyst are, in greater detail:

The preferred support used for preparation of the catalyst for pretreatment by 20 hydroisomerization-hydroconversion described within the scope of this patent is composed of silica SiO_2 and alumina Al_2O_3 . The silica content of the support, expressed as a percentage by weight, is generally between 1 and 90%, advantageously even between 5 and 95% and preferably between 10 and 80% and still more preferably between 20 and 70% and between 22 and 45%. This silica content is 25 perfectly measured by X-ray fluorescence.

For this particular type of reaction, the metal function is provided by a noble metal of Group VIII of the periodic table of the elements and more particularly platinum and/or palladium.

The noble metal content, expressed as a percentage by weight of metal relative 30 to catalyst, is generally between 0.05 and 10, and more preferably between 0.1 and 5.

The noble metal distribution represents the distribution of the metal inside the catalyst grain, the metal being able to be well or poorly dispersed. Thus, it is possible to obtain platinum which is poorly distributed (for example detected in a crown the thickness of which is clearly smaller than the grain radius) but well dispersed, i.e. all

the platinum atoms situated in the crown, will be accessible to the reagents. In our case, the distribution of the platinum is good, i.e. the profile of the platinum, measured according to the Castaing microprobe process, has a distribution coefficient above 0.1 and preferably above 0.2.

5 The BET surface area of the support is between $100\text{m}^2/\text{g}$ and $500\text{m}^2/\text{g}$ and preferably between $250\text{m}^2/\text{g}$ and $450\text{m}^2/\text{g}$ and for the silica-alumina-based supports, still more preferably between $510\text{m}^2/\text{g}$ and $450\text{m}^2/\text{g}$.

The preparation and shaping of the support, and in particular of the silica-alumina is done by usual methods known to a person skilled in the art.

10 Advantageously, prior to the impregnation of the metal, the support can undergo a calcination such as for example a heat treatment at $300\text{-}750^\circ\text{C}$ (600°C preferred) for 0.25-10 hours (2 hours preferred) under 0-30% volume of water vapour (for silica-alumina 7.5% preferred).

15 The noble metal salt is introduced by one of the usual methods used to apply the metal (preferably platinum and/or palladium, platinum being further preferred) to the surface of a support. One of the preferred methods is dry impregnation which consists of the introduction of the metal salt into a volume of solution which is equal to the pore volume of the mass of catalyst to be impregnated. Before the reduction operation, the catalyst can undergo a calcination such as for example a treatment 20 under dry air at $300\text{-}750^\circ\text{C}$ (520°C preferred) for 0.25-10 hours (2 hours preferred).

25 Before use in the hydroisomerization-conversion reaction, the metal contained in the catalyst must be reduced. One of the preferred methods of carrying out the metal reduction is treatment under hydrogen at a temperature between 150 and 160°C and a total pressure between 0.1 and 25 MPa. For example a reduction consists of a 2-hour plateau at 150°C then a rise in temperature to 450°C at a rate of $1^\circ\text{C}/\text{minute}$ then a 2-hour plateau at 450°C : throughout this reduction stage, the hydrogen flow rate is 1000 litres of hydrogen per litre of catalyst. It should also be noted that any *ex situ* reduction method is suitable.

30 The operating conditions under which the hydroisomerization-hydroconversion stage is carried out are important.

Pressure will be maintained between 2 and 25 MPa and preferably between 3 and 20 MPa and advantageously between 2 and 18 MPa, the space velocity is between 0.1 h^{-1} and 10 h^{-1} , preferably between 0.2 h^{-1} and 10 h^{-1} , and advantageously between 0.5 h^{-1} and 5.0 h^{-1} . The hydrogen level is between 100 and 2000 litres of hydrogen per

litre of charge and preferably between 150 and 1500 litres of hydrogen per litre of charge.

5 The temperature used in this stage is between 200 and 450°C and preferably between 250°C and 450°C, advantageously between 300°C and 450°C, and still more advantageously above 340°C, for example between 320 and 450°C.

In the case where a hydrorefining stage precedes the hydroisomerization-conversion stage, the two hydrorefining and stages of hydroisomerization-hydroconversion can be carried out on the two types of catalysts in (two or more) different reactors, and/or on at least two catalytic beds installed in the same reactor.

10 The use of the catalyst described above in the hydroisomerization-hydroconversion stage has the effect of increasing the isomerization rate of the heavy fraction (340°C⁺, or also 370°C⁺ and better still 380°C⁺), reducing its pour point and increasing its VI. More generally, it is found that the treatment of the hydroisomerization-hydroconversion stage then makes it possible to obtain better 15 yields of dewaxed oil fraction which will be obtained in the catalytic dewaxing stage.

20 In one embodiment, all of the effluent from the hydroisomerization-hydroconversion stage can be treated in the dewaxing process according to the invention. The embodiment, with passage into the catalytic dewaxing stage of all of the effluent from the hydroisomerization-hydroconversion stage, is economically beneficial, since one single distillation unit is used at the end of the process. Moreover, at the final distillation (after catalytic dewaxing or subsequent treatments) 25 a gasoil for very cold weather is obtained.

25 In a variant, the effluent from the hydroisomerization-conversion stage can undergo a separation of at least part (and preferably at least one major part) of light gases which include hydrogen and optionally also hydrocarbon compounds with a maximum of 4 carbon atoms. The hydrogen can be separated beforehand.

30 Advantageously, in another embodiment, the effluent from the hydroisomerization-conversion stage is distilled in order to separate light gases and also separate at least one residue containing compounds having a boiling point above at least 340°C. This is preferably an atmospheric distillation.

Distillation can advantageously be carried out in order to obtain several fractions (gasoline, kerosene, gasoil for example) having a boiling point of at most 340°C and a fraction (called residue) having an initial boiling point above at least 340°C and still better 350°C and preferably at least 370°C or 380°C.

According to a preferred variant of the invention, this fraction (residue) will then be treated in the catalytic dewaxing stage, i.e. without undergoing vacuum distillation. But in another variant, a vacuum distillation can be used.

Generally, in this text the term middle distillates refers to fraction(s) with an 5 initial boiling point of at least 150°C and final boiling point, ranging to just before the residue, i.e. generally up to 340°C, 350°C or preferably below 370°C or 380°C.

The effluent from the hydroisomerization-hydroconversion stage, can undergo, before or after distillation, other treatments such as for example extraction of at least some of the aromatic compounds.

10 At least part of the effluent from the hydroisomerization-hydroconversion stage, which effluent has optionally undergone the separations and/or treatments described above, is then subjected to the catalytic dewaxing process according to the invention.

15 It should be noted that compounds boiling above at least 340°C are always subjected to the catalytic dewaxing.

The effluent on leaving the catalytic hydrodewaxing process according to the invention is advantageously sent into the distillation train, which preferably integrates an atmospheric distillation and vacuum distillation section, which serves to separate conversion products with a boiling point below 340°C and preferably below 370°C, 20 (and including in particular those formed during the catalytic dewaxing stage), and separate the fraction which constitutes the oil base and the initial boiling point of which is above at least 340°C and preferably above or equal to 370°C.

Moreover, this vacuum distillation section makes it possible to separate the different grades of oil.

25 Preferably, before being distilled, the effluent leaving the catalytic hydrodewaxing process according to the invention is, at least partly and preferably wholly, sent onto a hydrofinishing catalyst in the presence of hydrogen so as to realize accelerated hydrogenation of the aromatic compounds that may still be present which impair the stability of the oils and distillates. However, the acidity of the catalyst must 30 be weak enough not to lead to the formation of cracking product with a boiling point below 340°C so as not to degrade the final yields in particular of oils.

The catalyst used in this hydrofinishing stage comprises at least one metal of Group VIII and/or at least one element of Group VIB of the periodic table. The strong metal functions: platinum and/or palladium, or nickel-tungsten, nickel-molybdenum

combinations, will be advantageously used in order to carry out accelerated hydrogenation of the aromatics.

This metals are deposited and dispersed on an amorphous or crystalline oxide-type support, such as for example, aluminas, silicas, silica-aluminas.

5 The hydrofinishing (HDF) catalyst can also contain at least one element of Group VII A of the periodic table of the elements. Preferably these catalysts contain fluorine and/or chlorine.

10 The contents by weight of metals are between 10 and 30% in the case of non-noble metals and below 2%, preferably between 0.1 and 1.5% and still more preferably between 0.1 and 1.0% in the case of noble metals.

The total halogen quantity is between 0.02 and 30% by weight, advantageously between 0.1 and 15%, or still more advantageously between 0.1 and 10%, preferably 0.01 to 5%.

15 Among the catalysts which can be used in this hydrofinishing stage, and leading to excellent performance values, and in particular in order to obtain medicinal oils, there can be mentioned catalysts containing at least one noble metal of Group VIII (platinum and VIII for example) and at least one halogen (chlorine and/or fluorine), the combination of chlorine and fluorine being preferred.

20 The operating conditions under which the hydrofinishing stage optionally following the catalytic dewaxing process according to the invention is carried out are the following:

- the reaction temperature is between 180 and 400°C and preferably between 210 and 350°C, advantageously 230-320°C;
- the pressure is between 0.1 and 25 MPa and preferably between 1.0 and 20 MPa;
- the hourly volume rate (hvr expressed as volume of charge injected per volume unit of catalyst per hour) is between approximately 0.05 and approximately 100 and preferably between approximately 0.1 and approximately 30 h⁻¹.

30 The contact between the charge and the catalyst is carried out in the presence of hydrogen. The amount of hydrogen used and expressed in litres of hydrogen per litre of charge is between 50 and approximately 2000 litres of hydrogen per litre of charge and preferably between 100 and 1500 normal litres of hydrogen per litre of charge.

Advantageously, the temperature of the hydrofinishing (HDF) stage is below the temperature of the catalytic hydrodewaxing (CHDW) stage. The difference $T_{CHDW} - T_{HDF}$ is generally between 20 and 200°C, and preferably between 30 and 100°C.

The effluent leaving the HDF section is then sent into the distillation train.

5 In the variant of the process according to the invention including a preliminary hydroisomerization-hydroconversion stage, the oil bases obtained in a process as described above have a pour point below -10°C, a VI above 95, preferably above 110 and still more preferably above 120, a viscosity of at least 3.0 cSt at 100°C, an ASTM colour below 1 and a UV stability such that the increase in the ASTM colour is
10 between 0 and 4 and preferably between 0.5 and 2.5.

The UV stability test, adapted from ASTM methods D925-55 and D1148-55, provides a quick method of comparing the stability of the lubricating oils exposed to a source of ultraviolet radiation. The test chamber is constituted by a metal enclosure equipped with a turntable which receives the oil samples. A bulb producing the same
15 ultraviolet rays as those of sunlight and placed at the top of the test chamber is directed downwards onto the samples. Among the samples is included a standard oil with known UV characteristics. The ASTM D1500 colour of the samples is determined at $t = 0$ then after 45 hours of exposure at 55°C. The results are transcribed for the standard sample and the test samples as follows:

20 a) initial ASTM D1500 colour,
b) final ASTM D1500 colour,
c) increase in colour,
d) cloudiness,
e) precipitate.

25 Another advantage of this variant of the process according to the invention is that it is possible to achieve very low aromatics contents, below 2% by weight, preferably 1% by weight and, better, below 0.05% by weight) and even to produce white oils of medicinal quality having aromatics contents below 0.01% by weight. These oils have UV absorbance values of 275, 295 and 300 nanometres respectively
30 of less than 0.8, 0.4 and 0.3 (ASTM method D2008) and a Saybolt colour between 0 and 30.

In particularly interesting fashion therefore, the process according to the invention also makes it possible to obtain medicinal white oils. Medicinal white oils are mineral oils obtained by accelerated refining of petroleum, their quality is subject

to different regulations aimed at guaranteeing their harmlessness for pharmaceutical applications, they are non-toxic and are characterized by their density and their viscosity. Medicinal white oils essentially comprise saturated hydrocarbons, they are chemically inert and their aromatic hydrocarbons content is low. Particular attention is 5 given to the aromatic compounds and in particular those with 6 polycyclic aromatic hydrocarbons (English abbreviation: PAHs) which are toxic and present in concentrations of one part per billion by weight of aromatic compounds in white oil. The total aromatic contents can be monitored using ASTM method D 2008, this UV adsorption test at 275, 292 and 300 nanometres makes it possible to monitor an 10 absorbance below respectively 0.8, 0.4 and 0.3 (i.e. the white oils have aromatics contents below 0.01% by weight). These measurements are carried out with concentrations of 1 g of oil per litre, in a 1 cm container. Marketed white oils differ in terms of their viscosity but also their original crude oil which can be paraffinic or naphthenic, these two parameters will cause differences both in the physico-chemical 15 properties of the white oils in question but also in their chemical composition.

The examples which follow illustrate the invention without however limiting its scope.

Example 1: Preparation of catalyst C1 according to the invention

20 Catalyst C1 comprises a ZBM-30 zeolite: This catalyst is obtained according to the procedure described below.

The ZBM-30 zeolite is synthesized according to BASF patent EP-A-46504 with the organic structuring agent triethylenetetramine.

25 The synthesized crude ZBM-30 zeolite is subjected to a calcination at 550°C under a dry air flow for 12 hours.

The thus-obtained H-ZBM-30 zeolite (acid form) has an Si/Al ratio of 45.

30 The zeolite is mixed with an SB3-type alumina gel supplied by Condéa. The mixed paste is then extruded through a die with a diameter of 1.4 mm. The thus-obtained extrudates are calcined at 500°C for 2 hours under air. The H-ZBM-30 content by weight is 80% by weight.

Then, the support extrudates are subjected to a stage of dry impregnation by an aqueous solution of platinum salt $\text{Pt}(\text{NH}_3)_4^{2+}$, 20H^- . The platinum content by weight of the thus-obtained catalyst C1 is 0.52%.

Example 2: Preparation of catalyst C2 not according to the invention

Catalyst C2 comprises a ZBM-30 zeolite. This catalyst is obtained according to the procedure described below.

The ZBM-30 zeolite is synthesized according to BASF patent EP-A-46504
5 with the organic structuring agent hexamethylenetetramine.

The crude synthesized ZBM-30 zeolite is subjected to a calcination at 550°C under a dry air flow for 12 hours.

The thus-obtained H-ZBM-30 zeolite (acid form) has an Si/Al ratio of 54.

The zeolite is mixed with an SB3-type alumina gel supplied by Condéa. The
10 mixed paste is then extruded through a die with a diameter of 1.4 mm. The thus-obtained extrudates are calcined at 500°C for 2 hours under air. The H-ZBM-30 content by weight is 80% by weight.

Then, the support extrudates are subjected to a stage of dry impregnation by an aqueous solution of platinum salt $\text{Pt}(\text{NH}_3)_4^{2+}$, 20H^- . The platinum content by weight
15 of the thus-obtained catalyst C2 is 0.53%.

Example 3: Preparation of catalyst C3 not according to the invention

Catalyst C3 comprises a ZSM-48 zeolite. This catalyst is obtained according to the procedure described in volume 33 pages 282-284 of the series Studies in
20 Surface Science and Catalysts. The zeolite is then subjected to calcination under a dry air flow for 12 hours at 550°C so as to eliminate the organic structuring agent used for its synthesis. The crude synthesized ZBM-48 zeolite is subjected to a calcination at 550°C under a dry air flow for 12 hours. The solid ZSM-48 obtained is subjected to 4 successive ion exchanges in a solution of 10N NH_4NO_3 , at approximately 100°C for 4
25 hours for each exchange.

The thus-obtained H-ZSM-48 zeolite (acid form) has an Si/Al ratio of 33.

The zeolite is mixed with an SB3-type alumina gel supplied by Condéa. The mixed paste is then extruded through a die with a diameter of 1.4 mm. The thus-obtained extrudates are calcined at 500°C for 2 hours under air. The H-ZSM-48 content by weight is 80% by weight.
30

Then, the support extrudates are subjected to a stage of dry impregnation by an aqueous solution of platinum salt $\text{Pt}(\text{NH}_3)_4^{2+}$, 20H^- . The platinum content by weight of the thus-obtained catalyst C3 is 0.50%.

Example 4: Use of catalysts C1 (according to the invention) and C2 (not according to the invention) for improving the pour point of a hydrocracking residue

Catalysts C1 and C2 the preparations of which are respectively described in

5 Examples 1 and 2 are used for preparing an oil base from the charge described in Table 1 below which is a hydrocracking residue from a vacuum distillate.

Table 1
Characteristics of the charge

Sulphur content (ppm by weight)	8
Nitrogen content (ppm by weight)	1
Pour point (°C)	+40
Initial point	281
10%	348
50%	418
90%	505
Final point	583

5 The catalysts are reduced beforehand under hydrogen at 450°C before the in situ catalytic test in the reactor. This reduction is carried out with plateaux. It consists of a 2-hour plateau at 150°C then a rise in temperature to 450°C at a rate of 1°C/minute then a 2-hour plateau at 450°C. During this reduction protocol, the hydrogen flow rate is 1000 litres of hydrogen per litre of catalyst.

10 The reaction takes place at 330°C, under a total pressure of 12 MPa, an hourly volume rate of 1 h⁻¹ and a hydrogen flow rate of 1000 litres of hydrogen per litre of charge.

The characteristics of the oil obtained, 380°C⁺ fraction, are shown in Table 2 below.

15

Table 2

	Catalyst C1	Catalyst C2
Viscosity Index VI of the 380°C ⁺ fraction	131	129
Pour point of the 380°C ⁺ fraction (°C)	-20	-20
Oil yield (380°C ⁺) (% by weight)	80.6	79.2

These examples show the full benefit of using catalyst C1 according to the invention synthesized in the presence of triethylenetetramine, which makes it possible

to reduce the pour point of the initial charge, in this case a hydrocracking residue, while still obtaining a high viscosity index (VI).

5 **Example 4: Use of catalysts C1 (according to the invention) and C3 (not according to the invention) for improving the pour point of a hydrocracking residue**

Catalysts C1 and C3 the preparations of which are respectively described in Examples 1 and 3 are used for preparing an oil base from the charge described in the table below, which is a hydrocracking residue from a vacuum distillate.

10

Table 3
Characteristics of the charge

Sulphur content (ppm by weight)	8
Nitrogen content (ppm by weight)	1
Pour point (°C)	+40
Initial point	281
10%	348
50%	418
90%	505
Final point	583

15

The catalysts are reduced beforehand under hydrogen at 450°C before the in situ catalytic test in the reactor. This reduction is carried out with plateaux. It consists of a 2-hour plateau at 150°C then a rise in temperature to 450°C at a rate of 1°C/minute then a 2-hour plateau at 450°C. During this reduction protocol, the hydrogen flow rate is 1000 litres of hydrogen per litre of catalyst.

20

The reaction takes place at 330°C, under a total pressure of 12 MPa, an hourly volume rate of 1 h⁻¹ and a hydrogen flow rate of 1000 litres of hydrogen per litre of charge.

The characteristics of the oil obtained, 380°C⁺ fraction, are shown in Table 4 below.

25

Table 4

	Catalyst C1	Catalyst C2
Viscosity index VI of the 380°C ⁺ fraction	131	129
Pour point of the 380°C ⁺ fraction (°C)	-20	-19
Oil yield (380°C ⁺) (% by weight)	80.6	78.9

These examples show the full benefit of using catalyst C1 according to the invention based on ZBM-30 synthesized in the presence of triethylenetetramine, 5 which makes it possible to reduce the pour point of the initial charge, in this case a hydrocracking residue, while still obtaining a high viscosity index (VI).

Example 6: Preparation of catalyst C4 for pretreatment by hydroconversion-hydroisomerization of the charge subjected to dewaxing

10

Catalyst C4 for pretreatment by hydroconversion-hydroisomerization is prepared from a silica-alumina support used in the form of extrudates. It contains 40% by weight of silica (SiO₂) and 60% by weight of alumina (Al₂O₃). The silica-alumina before addition of the noble metal has a surface area of 332 m²/g and its total pore 15 volume is 0.82 ml/g.

The catalyst C4 is obtained after impregnation of the noble metal on the support. The platinum salt H₂PtCl₆ is dissolved in a volume of solution corresponding to the total pore volume to be impregnated. The solid is then calcined for 2 hours under air at 500°C. The platinum content is 0.48% by weight. Measured over the 20 catalyst, the BET surface area is equal to 310 m²/g. The dispersion of the platinum measured by H₂/O₂ titration is 75%.

Example 7: Use of catalyst C1 (according to the invention) for improving the pour point of a hydrocracking residue pretreated on catalyst C4.

25

The catalyst the preparation of which is described in Example 6 is used to pretreat a charge which is a previously hydrorefined VGO in order to obtain oil bases. In order to be able to directly use the pretreatment stage catalysts, the charge was

therefore hydrorefined beforehand and the chief characteristics obtained of the charge are shown in Table 5 below.

5 **Table 5**
Characteristics of the charge

Sulphur content (ppm by weight)	3
Nitrogen content (ppm by weight)	1
Distillation:	
- initial point	335°C
- 10% point	397°C
- 50% point	459°C
- 90% point	537°C
- final point	600°C
Pour point	+ 42°C
VI after dewaxing with the solvent methylisobutylketone (MIBK) (-20°C)	105

This charge was used in order to evaluate catalyst C4 and prepare the charge which will then be hydrodewaxed catalytically on the catalyst C1.

10 The catalytic test unit comprises a fixed-bed reactor, with upflow of the charge, into which is introduced 80 ml of catalyst C4. This catalyst is then subjected to a pure-hydrogen atmosphere at a pressure of 12 MPa in order to ensure the reduction of the platinum oxide to metal platinum then the charge is finally injected. The total pressure is 12 MPa, the hydrogen flow-rate is 1000 litres of gaseous
15 hydrogen per litre of charge injected, the hourly volume rate is 0.9 h⁻¹ and the reaction temperature 345°C. After reaction, the effluents are fractionated to light products (gasoline bp-150°C), middle distillates (150-380°C) and residue (380°C⁺):

20 The net conversion achieved under these conditions of the charge to products having boiling points below 380°C is 28% by weight. The thus-obtained VI of the 380°C⁺ fraction after dewaxing with the solvent (-20°C) by methylisobutylketone (MIBK) is 122, i.e. a 17-point increase in the VI compared with the initial charge. On

the other hand, the pour point of the 380°C⁺ fraction has fallen only slightly, settling, as it does, at a value of +39°C compared with +42°C initially.

The 250-380°C fraction which corresponds to a gasoil cut and which results from the pretreatment stage converting hydrocracking residue has a pour point of - 5 18°C and a cetane number of 57, which makes it an excellent gasoil.

The residue (380°C⁺ fraction) prepared above is then dewaxed and hydrofinished in a unit comprising two fixed-bed reactors with ascending circulation of the charge ("up-flow"),

80 ml of the hydrodewaxing catalyst C1 prepared in Example 1 is introduced 10 into the first reactor and a hydrofinishing catalyst containing 1% by weight Pt, 1% by weight F and 1% by weight Cl on the alumina is introduced into the second reactor.

The catalysts are then subjected to a pure-hydrogen atmosphere at a pressure of 12 MPa in order to ensure the reduction of the platinum oxide to metal platinum then the charge (380°C⁺ fraction produced in Example 4) is injected.

15 The total pressure is 12 MPa, the hydrogen flow-rate is 1000 litres of gaseous hydrogen per litre of charge injected, the hourly volume rate in the first reactor containing C4 is 1.1 h⁻¹ and the reaction temperature 330°C and in the second reactor, containing the hydrofinishing catalyst, the total pressure is the same, except for losses of head, the reaction temperature is 220°C and all of the effluent from the first reactor 20 circulates at a space velocity of 0.5h⁻¹.

After reaction, the effluents are fractionated to light products (gasoline lp-150°C), middle distillates (150-380°C) and residue (380°C⁺).

25 The effluent is recovered then distilled under vacuum. The yields of oil fraction (380°C⁺ fraction) as well as the characteristics of the 380°C⁺ residue are shown in Table 6 for the charge, for the 380°C⁺ fraction from the converting pretreatment stage and for the 380°C⁺ fraction from the catalytic dewaxing and hydrofinishing stage.

Table 6

	1 Charge = hydrocracking residue	2 Stage (a) (pretreatment stage)	3 Stage (b) hydrofinishing (catalytic hydrodewaxing + hydrofinishing)
Reaction temperature (°C)	/	345	325 220
Total pressure (bar)		120	120 120
Conversion to 380°C ⁺ (% by weight)	/	32	/
Sulphur (ppm by weight)	3	/	/
Nitrogen (ppm by weight)	1	/	/
380°C⁺ Fraction			
Pour point (°C)	+42	+39	-21
380°C⁺ Fraction after treatment	380°C⁺ Fraction of the hydrocracking residue dewaxed with solvent	380°C⁺ Fraction hydroisomerized and dewaxed with solvent	380°C⁺ Fraction hydroisomerized, catalytically dewaxed and hydrofinished
VI	105	122	121
Pour point (°C)	-20	-20	-22
Saybolt colour	/	/	+30
UV absorption (D2008)		/	
260-280 nm	/	/	0.0006
280-290 nm	/	/	0.0004
290-300 nm	/	/	0.0003
300-360 nm	/	/	0.0002
360-400 nm	/	/	<0.0001
300-330 nm	/	/	0.0002
330-350 nm	/	/	<0.0001

This example demonstrates that the combination of a stage of pretreatment by hydroconversion-hydroisomerization (stage a) and a stage of catalytic dewaxing followed by hydrofinishing, leads to high-quality products. In particular it shows that stage (a) makes it possible to increase the viscosity index of the oil fraction (380°C⁺) 5 from 105 to 122 without sufficiently reducing the pour point (Table 6, columns 1 and 2). This reduction is carried out during stage (b) on the catalytic dewaxing catalyst which makes it possible to obtain a pour point of -22°C as well as a Saybolt colour of +30 which gives the product the quality of medicinal oil (Table 6, columns 2 and 3).

The entire disclosure of all applications, patents and publications, cited herein 10 and of corresponding French application No. 03/03.530, filed March 24, 2003 is incorporated by reference herein.

The preceding examples can be repeated with similar success by substituting the generically or specifically described reactants and/or operating conditions of this invention for those used in the preceding examples.

15 From the foregoing description, one skilled in the art can easily ascertain the essential characteristics of this invention and, without departing from the spirit and scope thereof, can make various changes and modifications of the invention to adapt it to various usages and conditions.